

Two New Alkaloids; Kikemanine and the Morphinandienone-type Alkaloid, Pallidine, from *Corydalis* Species

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Summary The structure of the morphinandienone-type alkaloid, pallidine (IV), from *Corydalis pallida* var. *tenuis* Yatabe has been determined.

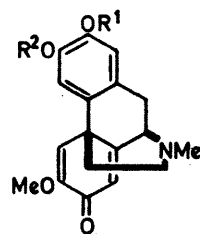
It has been suggested^{1,2} that flavinantine (I) and amurine (II) could be formed from reticuline (III) by intramolecular oxidative *para-para*-coupling to give morphinandienone (enantiomer of IV), followed by demethylation and re-methylation or by subsequent cyclisation of the resulting *O*-methyl-phenol. Supporting evidence came from the incorporation of (\pm)-reticuline (III) into flavinantine (I).³ In the hope of finding structures of type (IV), we studied the basic fractions of some *Papaveraceae*, and now report the presence of the morphinandienone (IV), "pallidine."

The basic extract from *Corydalis pallida* var. *tenuis* Yatabe, Japanese name, miyama kikeman (collected in Sendai in May) was separated by the usual method into non-phenolic bases [consisting of (-)-tetrahydropalmatine, protopine, and an unknown base, m.p. 159–160°, i.r. ν_{\max} (CHCl₃) 1692 and 1660 cm.⁻¹] and phenolic bases.†

The phenolic fraction consisted of capaurine, capaurimine, sinoacutine (V), isoboldine, and two new bases. One of the new alkaloids, named "kikemanine," C₂₀H₂₃NO₄ (VI), m.p. 177–178°, [α]_D¹⁸ - 287° (CHCl₃) was likely to correlate closely with alkaloid F-51,⁴ because it possessed three methoxy-groups and, on methylation with diazomethane, yielded (-)-tetrahydropalmatine. Furthermore, the mass spectrum revealed that the hydroxy-group was at C-9 or C-10.

The other alkaloid, pallidine, [α]_D¹⁸ - 32°, was assigned structure (IV) and the absolute configuration established on the basis of the following evidence. The i.r. (CHCl₃) of pallidine, ν_{\max} 3500, 1666, 1643, and 1624 cm.⁻¹, u.v. λ_{\max} (MeOH) 235 and 283 nm. (log ϵ 4.08 and 3.81), n.m.r. (τ in CDCl₃) 7.63 (3H; NMe), 6.21, 6.11 (3H, each; 2 \times OMe), 3.71 (1H; 8-H), 3.23 (1H; 5-H), 3.67 and 3.32 (1H each; 1-H and 4-H) and mass spectra (*m/e* 327, 312, 299) are consistent with those of a morphinandienone structure. The natural material proved to be identical with the so-called isosalutaridine synthesised previously by Franck⁵ and us,⁶ apart from optical activity, by full spectroscopic and chromatographic comparisons. The i.r. spectrum of *O*-methylpallidine (VII) was superimposable on that of *O*-methylflavinantine⁷ (VIII) which was derived from flavinantine (I). The o.r.d., [ϕ]₃₄₀ - 1250° (tr), [ϕ]₃₁₁ + 390° (pk), [ϕ]₂₈₇ - 9130° (tr), [ϕ]₂₇₀ - 2800° (pk),

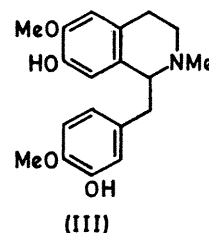
[ϕ]₂₅₄ - 12,770° (tr), and c.d. [Θ]₃₃₁ - 410°, [Θ]₂₉₈ + 4740°, [Θ]₂₈₀ - 3090°, [Θ]₂₆₄ + 1850°, curves of pallidine (IV) in methanol were similar to those obtained from sinoacutine⁸ (V) and furthermore, those of *O*-methyl compound (VII) and *O*-methylflavinantine (VIII) were entirely opposite. This is the first isolation of the morphinandienone-type alkaloids from *Corydalis* species. Pallidine (IV) is also the first in this group of alkaloids to correspond to an actual direct *p-p*-coupling product and is the enantiomer of the as yet unknown presumed intermediate in the *in vivo*



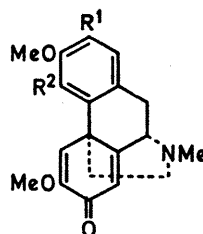
(I) R¹ = Me, R² = H

(II) R¹ + R² = CH₂

(VIII) R¹ = R² = Me



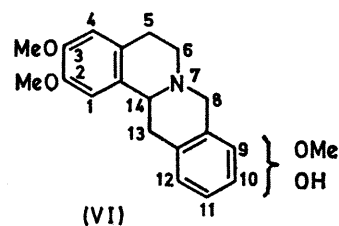
(III)



(IV) R¹ = OH, R² = H

(V) R¹ = H, R² = OH

(VII) R¹ = OMe, R² = H



(VI)

conversion of (III) into (I) and (II).

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The known alkaloids were characterised by full spectroscopic and physical comparisons with authentic samples.

¹ C. Chambers and K. L. Stuart, *Chem. Comm.*, 1968, 328.

² W. Döpke, H. Flentje, and P. W. Jeffs, *Tetrahedron*, 1968, **24**, 4459.

³ K. L. Stuart, V. Teetz, and B. Franck, *Chem. Comm.*, 1969, 333.

⁴ R. H. F. Manske, *Canad. J. Res.*, 1940, **B**, **18**, 80.

⁵ B. Franck, J. Lubs, and F. Dunkelmann, *Angew. Chem.*, 1967, **79**, 980.

⁶ T. Kametani, M. Koizumi, and K. Fukumoto, *Chem. and Pharm. Bull. (Japan)*, in the press, T. Kametani, K. Fukumoto, A. Kozuka, H. Yagi, and M. Koizumi, *J. Chem. Soc. (C)*, 1969, 2034.

⁷ T. Kametani, K. Fukumoto, F. Satoh, and H. Yagi, *J. Chem. Soc. (C)*, 1969, 520.

⁸ T. Kametani, M. Ihara, K. Fukumoto, and H. Yagi, *J. Chem. Soc. (C)*, 1969, 2030.